

A New Spiro-annulation Procedure: Intramolecular Decarboxylative Alkylation of β -Keto-esters

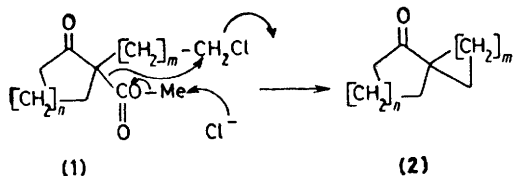
By ROBERT G. EILERMAN and BRIAN J. WILLIS*

*(Chemical Research Laboratories, Fritzsche Dodge and Olcott Inc., 76 Ninth Avenue,
New York, New York 10011)*

Summary A new intramolecular decarboxylative alkylation route to spirocyclic ketones, and its application to

the synthesis of (\pm)- β -vetivone and (\pm)- β -vetispirene are described.

ALTHOUGH many known methods of spirocyclization involve intramolecular alkylation routes,¹ there is need for an operationally simple procedure which avoids the use of a strong base. We report herein a regioselective intramolecular alkylation of enolates, generated *in situ* by halide-induced non-hydrolytic decarboxylation² of ω -halogeno- β -keto-esters (1), which provides a convenient route to spirocyclic ketones (2) (Scheme 1).



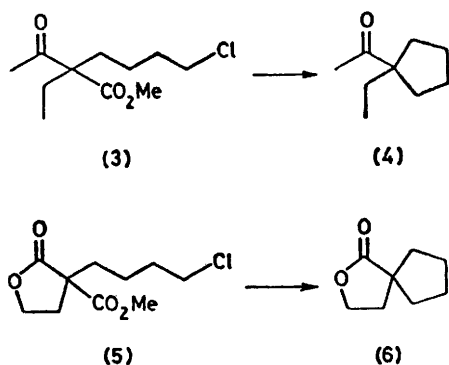
SCHEME 1.

Treating cycloalkanone carboxylates (1) with anhydrous lithium chloride (0.5 equiv.) in hexamethylphosphoramide at 125–140 °C provided the corresponding spiro-annulated ketones (2) in good yield. The results of representative experiments are summarized in the Table. Similar results were obtained with *N*-methyl-2-pyrrolidone as solvent, but dimethylformamide (DMF) was less satisfactory. Cyclization of the keto-ester (3) proceeded equally well, providing the acetylcyclopentane (4) in 80% yield. Similarly, the lactone carboxylate (5) gave the spiro-lactone (6) in 69% yield. This is believed to be the most direct general route to 2-spirolactones presently available.

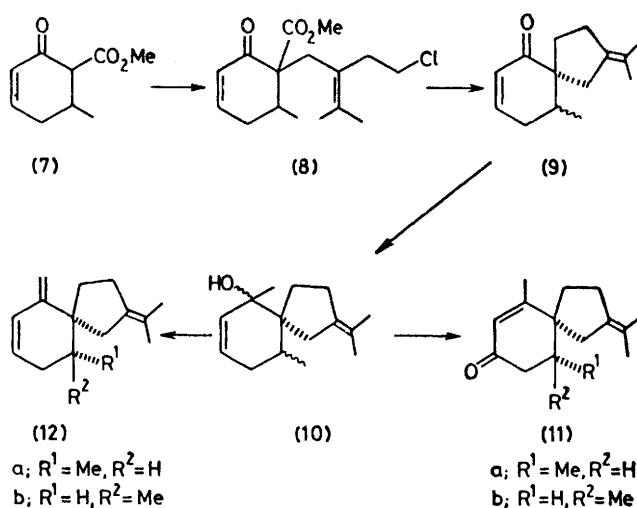
TABLE. Representative spirocyclization experiments.

| Compound (1) | Reaction time/h | Yield of (2)/% ^a |
|-------------------|-----------------|-----------------------------|
| a; $n = 1, m = 1$ | 1 | 64 |
| b; $n = 1, m = 3$ | 1 | 75 |
| c; $n = 2, m = 3$ | 1 | 68 |
| d; $n = 3, m = 3$ | 1.5 | 70 |
| e; $n = 8, m = 3$ | 1.5 | 71 ^b |

^a Yields reported are for isolated product. ^b M.p. 64–65 °C (lit.⁸ 50–52 °C).

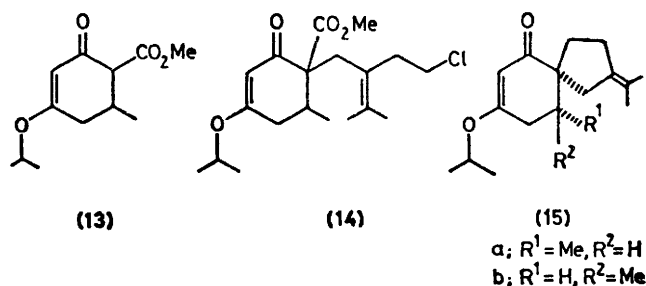


This spiro-annulation procedure has been successfully applied to the construction of the spiro[4.5]decane skeleton,³ resulting in a new synthesis of β -vetivone and β -vetispirene, components of the oil obtained by steam distillation of the roots of *Vetivera zizanioides*. Thus, reaction of the keto-ester (7)⁴ with 2-isopropylidene-1,4-dichlorobutane (sodium hydride-DMF, 3 h) afforded adduct (8) in 65% yield.† Decarboxylative spirocyclization of (8) gave the ketone (9) in 80% yield (9:1 mixture of epimers). Treatment of (9) with methyl-lithium afforded the allylic alcohol (10) which, without purification, was oxidized with pyridinium chlorochromate⁵ to give a mixture of (\pm)- β -vetivone (11a) and (\pm)-epi- β -vetivone (11b) (ratio 9:1) in 72% yield from (9). The alcohol (10) was also converted (toluene-*p*-sulphonic acid at 25 °C) into a mixture of (\pm)- β -vetispirene⁶ (12a) and its epimer (12b) (ratio 9:1) in 82% yield (Scheme 2).



SCHEME 2.

Application of the spirocyclization process to the Stork synthesis⁷ of β -vetivone resulted in an improved overall yield. The keto-ester (13) was treated with 2-isopropylidene-1,4-dichlorobutane (sodium hydride-DMF) to give (14) in 83% yield. Lithium chloride-induced decarboxy-



† All new compounds gave satisfactory i.r. and n.m.r. spectra, and microanalytical and/or mass spectral data in agreement with the assigned structures.

lation gave the spiroketone (**15**) (98%) as a 9:1 mixture of epimers, from which (**15a**) crystallized (m.p. 77—78 °C, from hexane) in 83% yield.† Treating (**15a**) with methyl-lithium, followed by hydrolysis with 1M hydrochloric acid, gave pure racemic β -vetivone (**11a**) (m.p. 49—50 °C, from hexane) in 87% yield.

(Received, 7th October 1980; Com. 1097.)

† The stereochemistry of (**15a**) was confirmed by analytical h.p.l.c. comparison with an authentic sample prepared by the method of Stork.⁷

¹ A. P. Krapcho, *Synthesis*, 1974, 383.

² For examples of simple alkylations using a similar technique see (a) M. Asaoka, K. Miyake, and H. Takei, *Chem. Lett.*, 1975, 1149, and (b) S. Takei and Y. Kawano, *Tetrahedron Lett.*, 1975, 4389.

³ J. A. Marshall, S. F. Brady, and N. H. Anderson, *Fortschr. Chem. Org. Naturst.*, 1974, **31**, 283.

⁴ F. Bohlmann and K. Prezewowsky, *Chem. Ber.*, 1964, **97**, 1176.

⁵ W. G. Dauben and D. M. Michno, *J. Org. Chem.*, 1977, **42**, 682.

⁶ N. H. Anderson, M. S. Falcone, and D. D. Syrdal, *Tetrahedron Lett.*, 1970, 1759.

⁷ G. Stork, R. L. Danheiser, and B. Ganem, *J. Am. Chem. Soc.*, 1973, **95**, 3414.

⁸ H. Nozaki, H. Yamamoto, and T. Mori, *Can. J. Chem.*, 1969, **47**, 1107.